

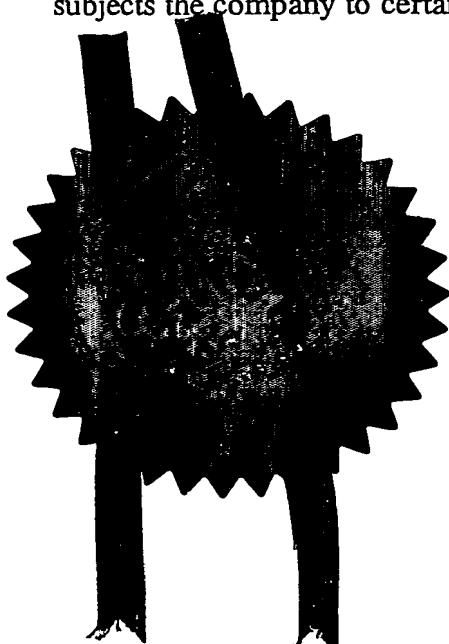
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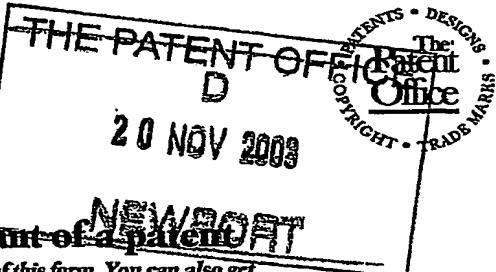


Evans.

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0327026.1

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1. Your reference

0327026.1

2. Patent application number

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3. Full name, address and postcode of the or of each applicant (underline all surnames)

INTELLIENT
 2nd FLOOR LPL
 145-147 ST JONNS STREET
 LONDON
 EC1V 4PY

Patents ADP number (if you know it)

8756835001

If the applicant is a corporate body, give the country/state of its incorporation

4. Title of the invention

VOLUME MEASURING DEVICE

5. Name of your agent (if you have one)

"Address for service" in the United Kingdom
 to which all correspondence should be sent
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Patents ADP number (if you know it)

6. Priority: Complete this section if you are declaring priority from one or more earlier patent applications, filed in the last 12 months.

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7. Divisionals, etc: Complete this section only if this application is a divisional application or resulted from an entitlement dispute (see note 8)

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Continuation sheets of this form	0 2810
Description	9 NINE
Claim(s)	0 2810
Abstract	0 2610
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DESCRIPTION

Volume Measuring Device

The invention relates to a volume measuring device designed to determine the separate volumes of the gaseous and non-gaseous fractions of a gaseous and non-gaseous mixture contained within a vessel.

An example of the application of this invention would be to measure the volume of gas above the liquid level in some liquid-containing vessel or the volume of gas within a container that contained bulk solids or even solid particulates such as dust and powder. The volume of the non-gaseous material would then be derived from simply knowing the overall volume of the containing vessel.

Currently used apparatus for measuring liquid levels are prone to inaccuracies that arise due to the position of the vessel with respect to the horizontal and also the irregular shape of many vessels.

Currently, the measurement of the solid phase within a vessel that contains a two phase mixture of that solid and a gaseous fraction is also very limited with present apparatus. Measurements of the amount of bulk solid material rely upon some estimation of the original volume of the solid within the vessel and an approximate record of the material that has been removed over time. Alternatively, the container is weighed, the tare weight removed and the weight/volume of the solid thereby calculated. There is currently no existing technique for the measurement of the volume of a dust or particulate solid that is air-borne within a vessel (e.g coal or flour dust etc.)

The invention is an improvement upon prior art as its essential features are dependent upon Boyle's Law. The instantaneous volume of the gaseous fraction contained within a vessel under test is changed by a very small amount. As a consequence of this change in volume, the gaseous pressure within the vessel under test changes in an inversely proportional fashion.

The ratio of the small change in volume to the overall instantaneous volume of the gaseous fraction within the vessel dictates the magnitude of the pressure change. Prior art relies upon this principle to determine the instantaneous volume of the gaseous fraction of material within the vessel under test, and by a process of subtraction, the volume of the remaining non gaseous fraction. This, however, requires the knowledge or measurement of the pressure, and possibly the temperature, of the gaseous fraction both before and after the change in gaseous volume.

This new invention relies upon the relationship between the amount of work required to effect small changes to the overall instantaneous volume of the gaseous fraction within the vessel. It is this relationship which allows for the determination of the volume of the gaseous fraction and thus the volume of the non gaseous fraction of material with the vessel. In so doing, it avoids the need for knowing the pressure or temperature.

The preferred embodiment of this invention is a linear solenoid-based piston-type device, although other configurations of actuator are possible.

Referring to the drawings, Figures 1, 3, 5, 7 represent four alternative but essentially identical manifestations of the invention. Figures 1, 3, 5 and 7 show the invention with the piston in a fully retracted position while Figures 2, 4, 6 and 8 show the corresponding manifestation of the invention with the piston in the fully extended position. Figure 9 illustrates the invention, ie the actuator, in relation to the vessel containing the gaseous and non-gaseous material to be measured. Figure 10 illustrates positions of the piston stroke where measurements are taken, albeit that measurements are recorded continually.

The manifestation of the invention illustrated by Figures 1 and 2 show:

- Labelled 1 - a position sensor, which measures the relative position of the piston, in the form of a LVDT (linear variable displacement transducer). It is possible to use a different form of position sensor.
- Labelled 2 - a motive solenoid which drives the piston.
- Labelled 3 - a vent to equalise pressure to external ambient pressure and thus minimise resistance
- Labelled 4 - the cylinder
- Labelled 5 - the piston and shaft with a flush fitting piston head, the shaft also forming the solenoid core.

The manifestation of the invention illustrated by Figures 3 and 4 show:

- Labelled 3 - a vent to equalise pressure to external ambient pressure
- Labelled 4 - the cylinder
- Labelled 5 - the piston and shaft with a flush fitting piston head
- Labelled 6 - a multi-tapped coil incorporating the motive solenoid and LVDT as either single or multiple windings. Again it is possible to use a different form of position sensor.

The manifestation of the invention illustrated by Figures 5 and 6 show:

- Labelled 1 - a position sensor, which measures the relative position of the piston, in the form of a LVDT (linear variable displacement transducer).
- Labelled 2 - a motive solenoid which drives the piston.
- Labelled 3 - a vent to equalise pressure to external ambient pressure
- Labelled 4 - the cylinder
- Labelled 7 - the piston and shaft with a non flush fitting piston head
- Labelled 8 - the flexible/elastic, air tight diaphragm operating within its normal elastic parameters

The manifestation of the invention illustrated by Figures 7 and 8 show:

- Labelled 3 - a vent to equalise pressure to external ambient pressure
- Labelled 4 - the cylinder
- Labelled 6 - a multi-tapped coil incorporating the motive solenoid and LVDT as either single or multiple windings.
- Labelled 7 - the piston and shaft with a non flush fitting piston head
- Labelled 8 - the flexible/elastic, air tight diaphragm operating within its normal elastic parameters

The preferred manifestation of the invention will be determined by the application environment and costs associated with the required manufacturing tolerances.

Figure 9 shows:

- Labelled 9 – the invention/actuator in relation to the vessel containing the gaseous and non-gaseous fractions to be measured
- Labelled 10 - the vessel containing the gaseous and non-gaseous fractions to be measured
- Labelled 11 - the non-gaseous fraction within the vessel
- Labelled 12 – the initial volume of gaseous fraction within the vessel prior to any change being introduced by the invention/actuator. This includes the maximum actuator induced volume change labelled 13.
- Labelled 13 – the maximum invention/actuator induced volume change where this volume change is no more than 0.1% of the overall volume of the vessel labelled 10. This ensures an accuracy of at least 1%.

Figure 10 illustrates the invention introducing a change in volume of the gaseous fraction of material within the vessel.

- Labelled 14 – The starting position of a piston stroke.
- Labelled 15 – The linear displacement of the piston from position labelled 14 at some point through a compression stroke.
- Labelled 17 – The volume change associated with the piston moving from positions labelled 14 to 15 ie the Cross-sectional area of the cylinder (4) x the linear displacement between positions labelled 14 to 15.
- Labelled 16 – The linear displacement of the piston from position labelled 14 at some point through a compression stroke.
- Labelled 18 – The volume change associated with the piston moving from positions labelled 15 to 16 ie the Cross-sectional area of the cylinder (4) x the linear displacement between positions labelled 15 to 16.

There are 3 method by which the invention achieves the required goal of determining the separate volumes of the gaseous and non gaseous fractions within a vessel. The manifestation of the invention illustrated in Figure 3 and 4 will be used for the rest of this description. However, any of the manifestations could be used.

The invention is attached to the test vessel containing the gaseous and non gaseous mixture as shown in Figure 9. Either of the 3 separate method described below can then be used to determine the volume of the gaseous fraction of material within the vessel. Then, by simple subtraction of the gaseous volume from overall volume of the vessel, one can determine the volume of the non-gaseous fraction of material within the vessel. This information is then displayed on some convenient external device.



Method 1:

An electric current is applied to the motive solenoid (6) causing the piston to move quickly and suddenly along its full stroke. This causes the volume of the gaseous fraction within the vessel to change by a very small amount, very rapidly. The high speed by which this change occurs is important so as to prevent leakage of the gaseous fraction out of the vessel thus contravening Boyle's Law.

The position sensing LVDT (6) is able to determine the exact linear displacement of the piston throughout its stroke. By knowing the cross-sectional area of the cylinder (4) and the linear displacement it is possible to determine the incremental volume changes introduced by the invention

ie

$$v = \text{Cross-sectional area of the cylinder} \times \text{linear displacement}$$

Applying Boyle's Law

$$P_0 V = P(V-v)$$

where P_0 – initial gaseous pressure

V – initial gaseous fraction volume

P – gaseous pressure after volume change v

v – small change in volume

so

$$P = \frac{P_0 V}{(V-v)} = P_0 \left(1 - \frac{v}{V}\right)^{-1}$$

then the rate of change of the pressure of the gaseous fraction with incremental changes in volume v , ie the 1st differential, is

$$\frac{dP}{dv} = P' = -P_0 \left(1 - \frac{v}{V}\right)^{-2} * \frac{-1}{V} = \frac{P_0}{V} \left(1 - \frac{v}{V}\right)^{-2} \quad \text{Equation (1)}$$

and the rate of the rate of change of the pressure of the gaseous fraction with incremental changes in volume v , ie the 2nd differential, is:

$$\frac{d^2P}{dv^2} = P'' = \frac{-2P_0}{V} \left(1 - \frac{v}{V}\right)^{-3} * \frac{-1}{V} = \frac{2P_0}{V^2} \left(1 - \frac{v}{V}\right)^{-3} \quad \text{Equation (2)}$$

If one now divide the Equation (2) by Equation (1) ie. 2nd differential by the 1st differential,

$$\frac{P'}{P''} = \frac{P_0}{V \left(1 - \frac{v}{V}\right)^2} * \frac{V^2 \left(1 - \frac{v}{V}\right)^3}{2P_0}$$

it follows that

$$\frac{P'}{P''} = \frac{V}{2} * \left(1 - \frac{v}{V}\right) = \frac{-v}{2} + \frac{V}{2} \quad \text{Equation (3)}$$

ie. a classic $y = mx + c$ linear relationship between the changing pressure and small, incremental changes in volume v . The volume of the gaseous material within the vessel is therefore twice the value of the intercept of a "best fit" line (resulting from linear regression analysis) with the y -axis.

Now, the amount of electrical work done (W) to change the volume of the gaseous fraction of material in the vessel at a given instance can be shown to be:

$$W = \text{Force} \times \text{the linear displacement of the piston}$$

The resistive force offered against this volume change by the gaseous material can be shown to be:

$$\text{Force} = \text{Gaseous Pressure} \times \text{Cross-sectional area of the cylinder} \quad (4)$$

So, at any given instance it can be shown that the work done is proportional to the instantaneous pressure of the gaseous fraction within the vessel ie.

$$W \propto P$$

Hence, by substituting the rate of change of work done and the rate of the rate of change of work done into Equation (3), one obtains;

$$\frac{W'}{W''} = \frac{V}{2} * \left(1 - \frac{v}{V}\right) = \frac{-v}{2} + \frac{V}{2} \quad \text{Equation (4)}$$

The instantaneous voltage and current applied to the motive solenoid (6) and the linear displacement of the piston are measured continuously and integrated with respect to time, in order to determine the dynamic measure of the actual work consumed by the actuator as it proceeds through its stroke. The 1st and 2nd derivatives required by Equation (4) can be either supplied directly via analogue electronics (differentiators) or obtained by digital computation once the signals have been recorded via standard A/D conversion. Subsequent application of Equation (4) then allows the determination of the volume of the gaseous fraction of material within the vessel.

Method 2:

An electric current is applied to the motive solenoid (6) causing the piston to move quickly and suddenly along its full stroke. The current and voltage are exactly regulated so that the work done, ie voltage x current, is kept constant. This causes the volume of the gaseous fraction within the vessel to change by a very small amount, very rapidly. The high speed by which this change occurs is important so as to prevent leakage of the gaseous fraction out of the vessel thus contravening Boyle's Law.

The position sensing LVDT (6) is able to determine the exact linear displacement of the piston throughout its stroke. By knowing the cross-sectional area of the cylinder (4) and the linear displacement it is possible to determine the small volume changes (dv) introduced by the invention

ie

$$dv = \text{Cross-sectional area of the cylinder (4)} \times \text{linear displacement (d)}$$

Now, the amount of electrical work done (W) to change the volume of the gaseous fraction of material in the vessel at a given instance can be shown to be:

$$W = \text{Force} \times \text{the linear displacement of the piston}$$

$$\text{ie. } W = F \times d$$

The resistive force offered against this volume change by the gaseous material can be shown to be:

$$\text{Force} = \text{Pressure of the Gaseous fraction} \times \text{Cross-sectional area of the cylinder (4)}$$

Substituting with the volume change (dv), and applying Boyle's Law, ie $PV=k$, where k is some constant, to substitute for the pressure P, gives:

$$W = k \frac{1}{V} dv \quad \text{Equation (5)}$$

Therefore, the electrical work done (W_1) to introduce a small change in volume from V to $V-v_1$, where V is the initial gaseous fractional volume of the vessel, labelled 12 in Figure 9, and v_1 is the small volume change introduced, labelled 17 in Figure 10, can be shown to be :

$$W_1 = k \int_v^{V-v_1} \frac{1}{V} dv \quad \text{Equation (6)}$$

Performing this integration, gives

$$W_1 = k \ln \left| \frac{V - v_1}{V} \right| \quad \text{Equation (7)}$$

As the piston stroke proceeds, additional electrical work (W_2) is done and the gaseous fraction of the whole system is further reduced by a small volume v_2 , labelled 18 in Figure 10, from $V-v_1$ to $V-v_1-v_2$. Then, substituting these values into the functional relationship described by Equation (5), gives:

$$W_2 = k \ln \left| \frac{V - v_1 - v_2}{V - v_1} \right| \quad \text{Equation (8)}$$

Since the work done over successive fixed intervals of time is kept constant, $W_1 = W_2$. Therefore, equating Equation (7) to Equation (8), yields:

$$k \ln \left| \frac{V - v_1}{V} \right| = k \ln \left| \frac{V - v_1 - v_2}{V - v_1} \right|$$

so,

$$\frac{V - v_1}{V} = \frac{V - v_1 - v_2}{V - v_1}$$

and finally

$$V = \frac{v_1^2}{v_1 - v_2} \quad \text{Equation (9)}$$

The instantaneous voltage and current applied to the motive solenoid (6) are measured simultaneously with the instantaneous value of the linear displacement of the piston in order to monitor the change in volume of the gaseous fraction resulting from a specific amount work throughout the piston stroke. By equating the linear displacement to volume changes, (ie linear displacement x cross-sectional area of the cylinder (4)) and applying Equation (9) to the data recovered either by analogue electronic or computation, we can determine the volume of the gaseous fraction of material within the vessel.

Alternatively, since the electrical work is kept constant over successive fixed intervals of time, one could simply measure the change in the linear displacement of the piston at fixed intervals. It can be shown that the general series solution to Equation (9) that relates the total volume of the system V to the continuous process of n th instantaneous volume measurement (v) of the actuator as it proceeds through it cycle can be described by Equation (10) below :

$$V = \frac{v_{n-1}^2}{v_{n-1} - v_n} + \sum_{n=0}^{n-2} v_n \quad \text{Equation (10)}$$

Method 3:

This method uses the motive solenoid (6) simply to restore the piston (5) to its most retracted position after each compression stroke. The piston is move quickly and suddenly along its compression stroke by some constant force. The source of this force could be electromagnetic, spring, pneumatic, gravity or anything else provided that it is constant throughout the compression stroke of the piston.

This changes the volume of the gaseous fraction within the vessel to change by a very small amount, very rapidly. The high speed by which this change occurs is important so as to prevent leakage of the gaseous fraction out of the vessel thus contravening Boyle's Law. The rate of compression can be slower depending upon how efficiently the vessel is sealed against and gaseous leakage.

The position sensing LVDT (6) is able to determine the exact linear displacement of the piston throughout its stroke. By knowing the cross-sectional area of the cylinder (4) and the linear displacement it is possible to determine the small volume changes (dv) introduced by the invention

ie

$$dv = \text{Cross-sectional area of the cylinder (4)} \times \text{linear displacement (d)}$$

Now, the amount of work done (W) to change the volume of the gaseous fraction of material in the vessel at a given instance can be shown to be:

$$W = \text{Force} \times \text{the linear displacement of the piston}$$

$$\text{ie. } W = F \times d$$

The resistive force offered against this volume change by the gaseous material can be shown to be:

$$\text{Force} = \text{Pressure of the Gaseous fraction} \times \text{Cross-sectional area of the cylinder (4)}$$

Substituting with the volume change (dv), and applying Boyle's Law, ie $PV=k$, where k is some constant, to substitute for the pressure P , gives:

$$W = k \frac{1}{V} dv \quad \text{Equation (5) as in Method 2}$$

Therefore, the work done (W_1) to introduce a small change in volume from V to $V-v_1$ where, V is the initial gaseous fractional volume of the vessel, labelled 12 in Figure 9, and v_1 is the small volume change introduced, labelled 17 in Figure 10, can be shown to be :

$$W_1 = k \int_v^{V-v_1} \frac{1}{V} dv \quad \text{Equation (6) as in method 2}$$

Performing the integration, we gives:

$$W_1 = k \ln \left| \frac{V - v_1}{V} \right| \quad \text{Equation (7) as in method 2}$$

As the piston stroke continues, and more work (W_2) is done, the gaseous fraction is further reduced by a small volume v_2 , labelled 18 in Figure 10, from $V - v_1$ to $V - v_1 - v_2$. Then, substituting into Equation (5), gives:

$$W_2 = k \ln \left| \frac{V - v_1 - v_2}{V - v_1} \right| \quad \text{Equation (8) as in method 2}$$

Since the work done over successive fixed intervals of time is kept constant, $W_1 = W_2$. Therefore, equating Equation (7) to Equation (8), yields:

$$k \ln \left| \frac{V - v_1}{V} \right| = k \ln \left| \frac{V - v_1 - v_2}{V - v_1} \right|$$

so,

$$\frac{V - v_1}{V} = \frac{V - v_1 - v_2}{V - v_1}$$

and finally

$$V = \frac{v_1^2}{v_1 - v_2} \quad \text{Equation (9) as in method 2}$$

Since the work done in compressing the gaseous fraction is constant, then the work done in any unit period of time is also constant. The instantaneous value of the linear displacement of the piston is measured in order to monitor the change in volume of the gaseous fraction for a specific amount of time throughout the piston stroke. By equating the linear displacement to volume changes, (ie linear displacement \times cross-sectional area of the cylinder (4)) and applying Equation (9) to the data recovered either by analogue electronic or computation, we can determine the volume of the gaseous fraction of material within the vessel.

Alternatively, since the work is kept constant over successive fixed intervals of time, one could simply measure the change in the linear displacement of the piston at fixed intervals. It can be shown that the general series solution to Equation (9) that relates the total volume of the system V to the continuous process of n th instantaneous volume measurement (v) of the actuator as it proceeds through it cycle can be described by Equation (10) below :

$$V = \frac{v_{n-1}^2}{v_{n-1} - v_n} + \sum_{n=0}^{n-2} v_n \quad \text{Equation (10) as in method 2}$$

Drawings
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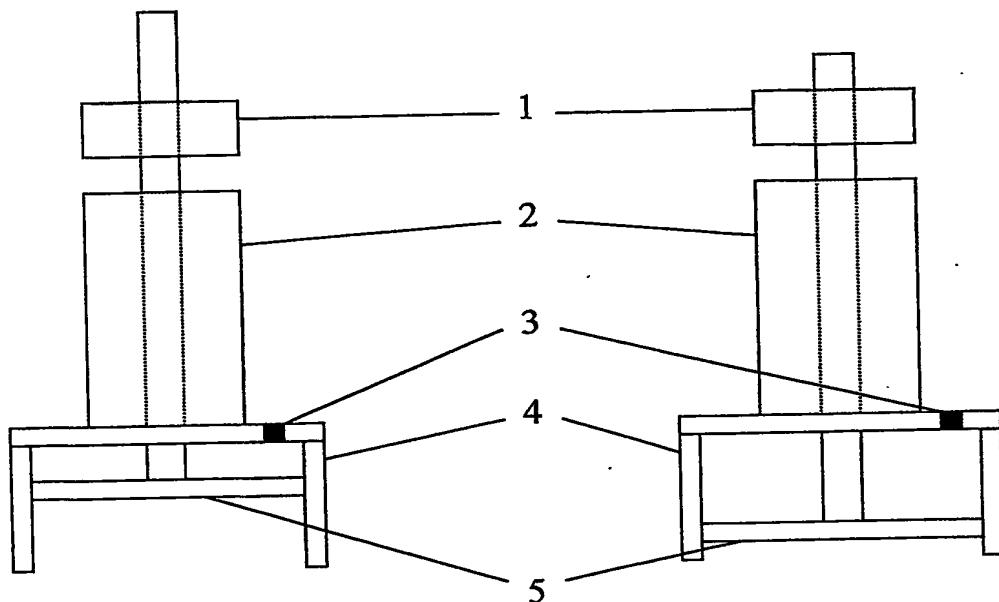


Figure 1

Figure 2

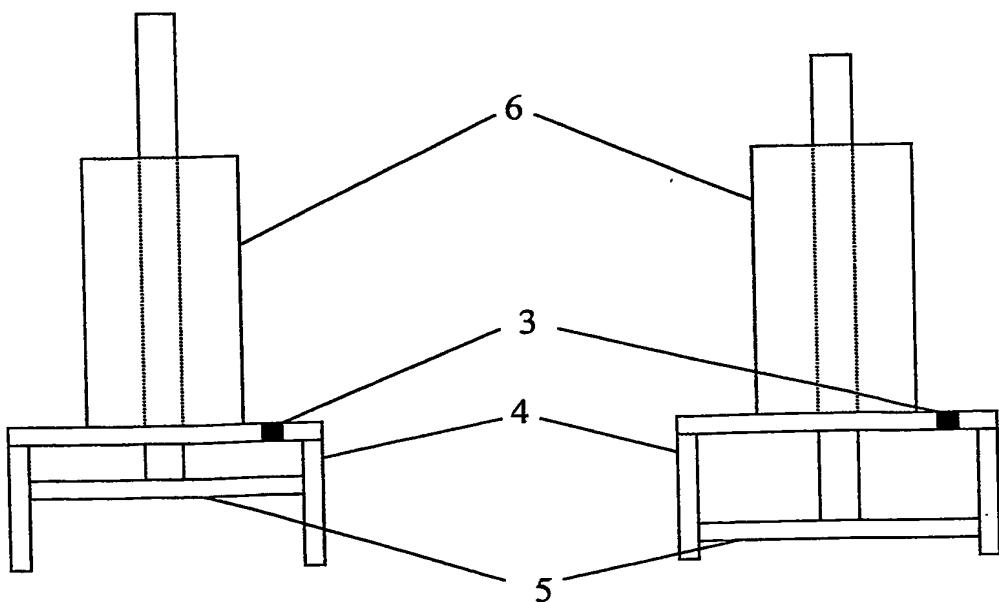
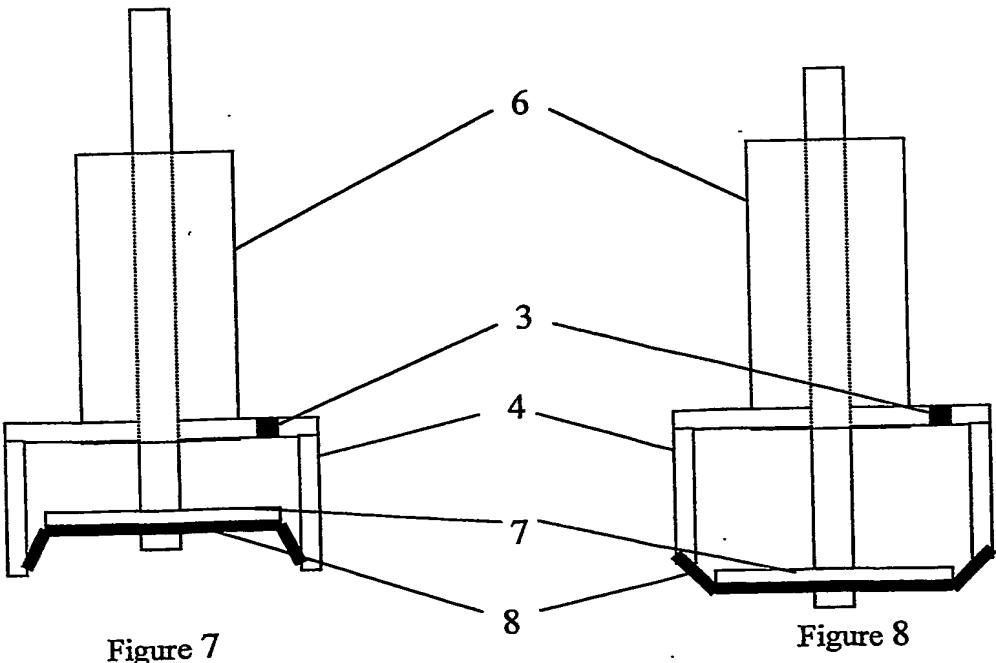
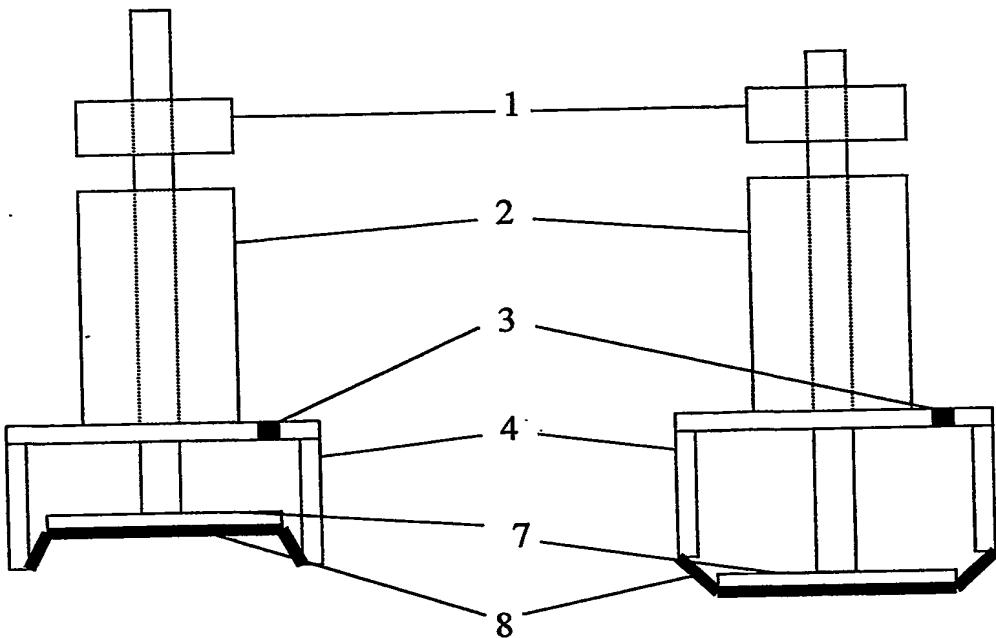


Figure 3

Figure 4

Drawings
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Drawings
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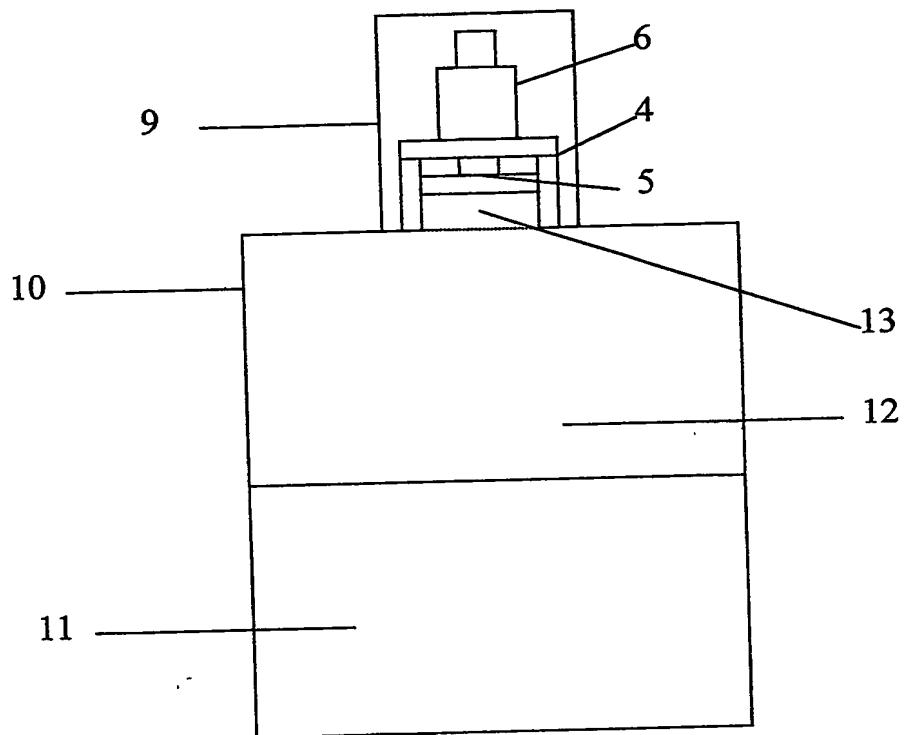


Figure 9

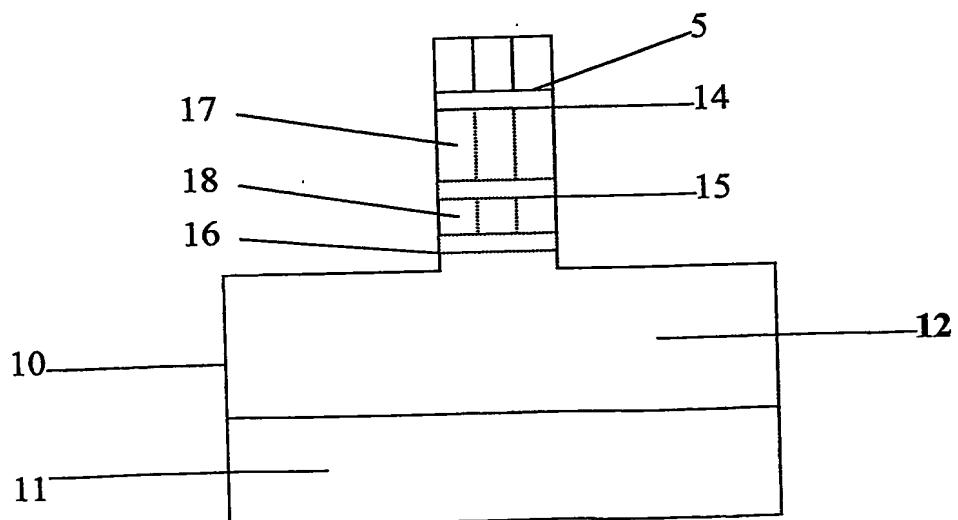


Figure 10

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